

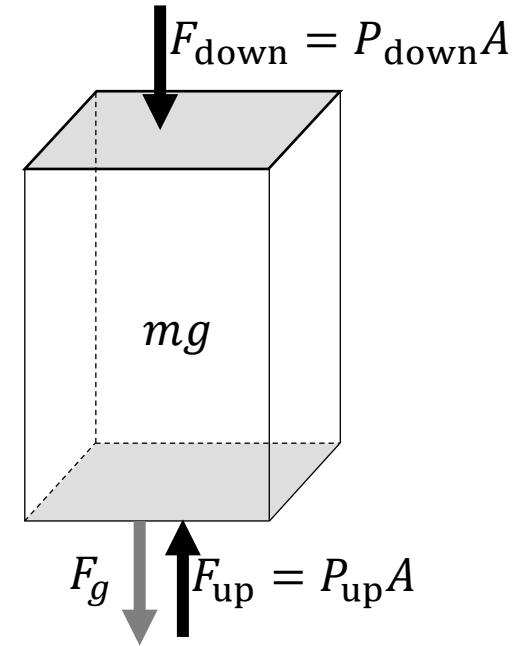
Quick notes on
Simple Atmospheres

Zach Meisel

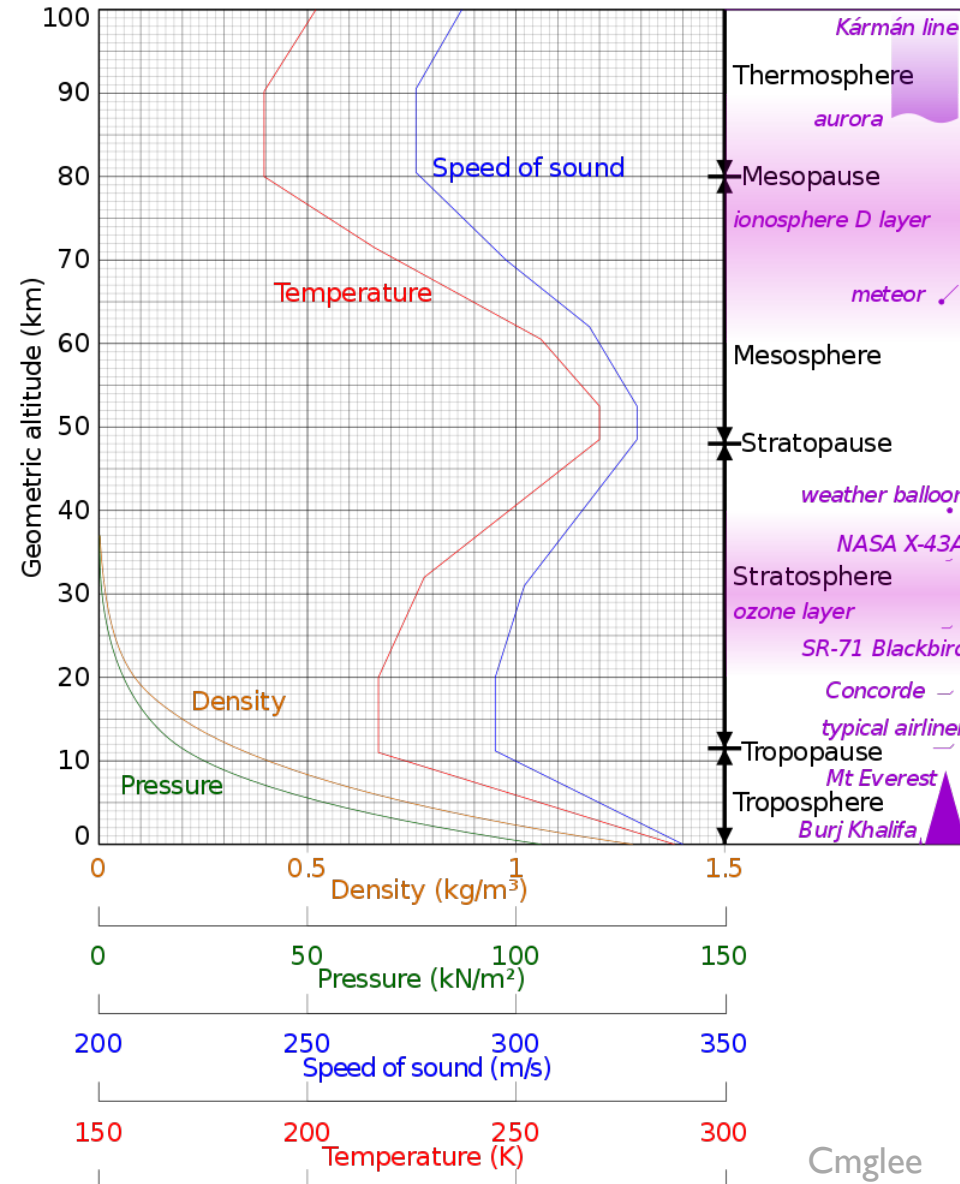
Ohio University - ASTR4201 - Fall 2020

the Isothermal Atmosphere

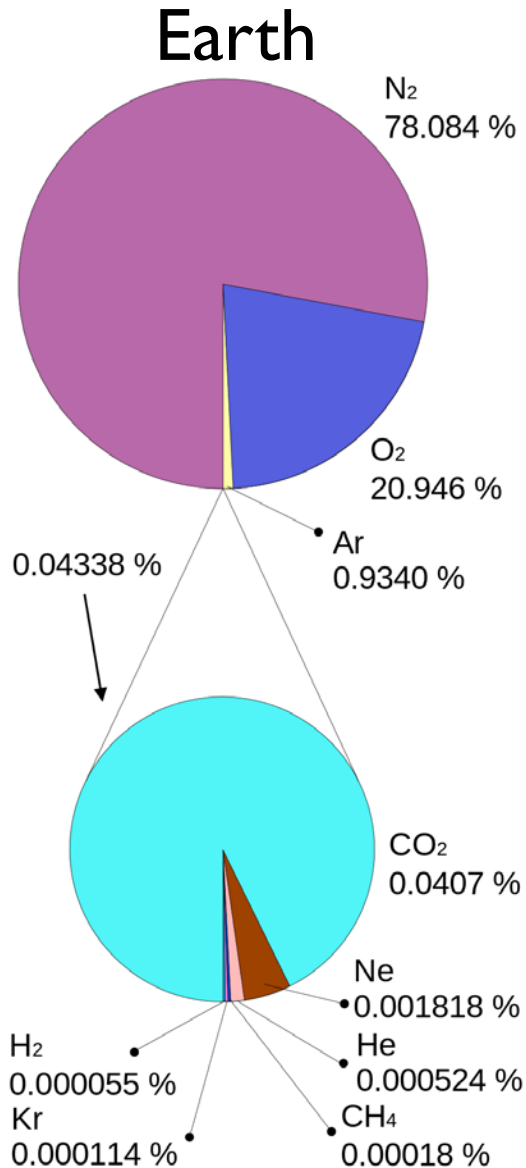
- Hydrostatic equilibrium:
 - Must satisfy $\vec{F}_{net} = m\vec{a} = 0$ for a static fluid element
 - Forces: Gravity downward & pressure inward from all sides
 - Non-vertical pressure forces cancel
 - Vertical forces cancel: $F_{\text{pressure,up}} = F_{\text{pressure,down}} + F_{\text{gravity}}$
 - $F_{\text{pressure}} = P \cdot A$
 - $A(P_{\text{up}} - P_{\text{down}}) = F_g = mg = \rho Vg = \rho \Delta x Ag$
 - $-\frac{\Delta P}{\Delta x} = \rho g \rightarrow \frac{dP}{dr} = -\rho g(r)$
- Equation of state:
 - $PV = Nk_B T \rightarrow P = \frac{N}{V} k_B T = \frac{k_B}{\mu m_u} \rho T \rightarrow \rho = \frac{\mu m_u}{k_B T} P$
- Substitute $\rho(P)$ into hydrostatic equilibrium equation and solve:
 - $P(r) = P_{\text{ref}} \exp(-r/H_P)$, where H_P is the pressure scale height
This is the “barometric formula”



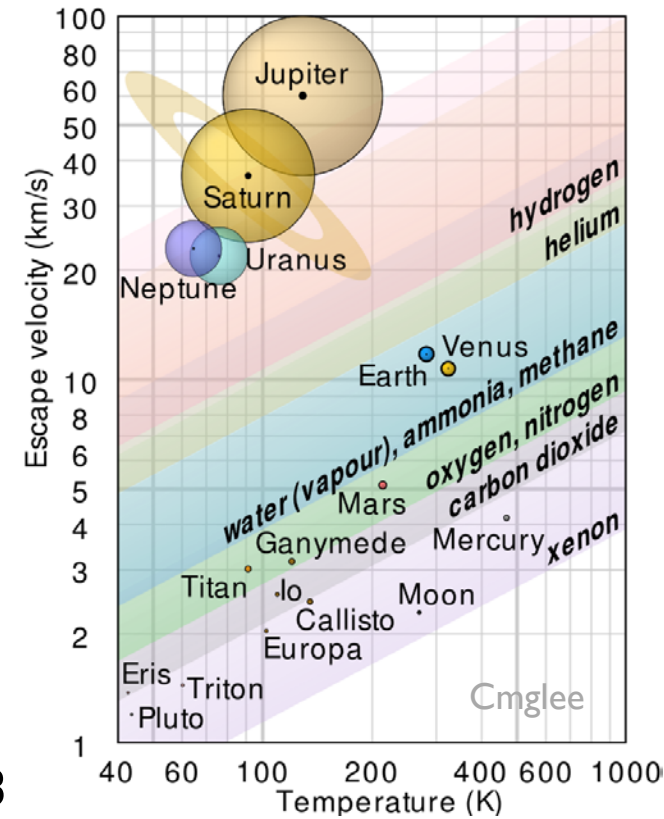
the Isothermal Atmosphere



Atmosphere Composition (for μ)



- Venus: ~97% CO₂, ~3% N₂
- Mars: ~96% CO₂, ~2% Ar, ~2% N₂
- Jupiter: ~90% H₂, ~10% He
- Saturn: ~96% H₂, ~4% He
- Uranus: ~83% H₂, ~15% He, ~2% CH₄
- Neptune: ~80% H₂, ~20% He



Pro-solar abundances calculated from Lodders *ApJ* 2003

