

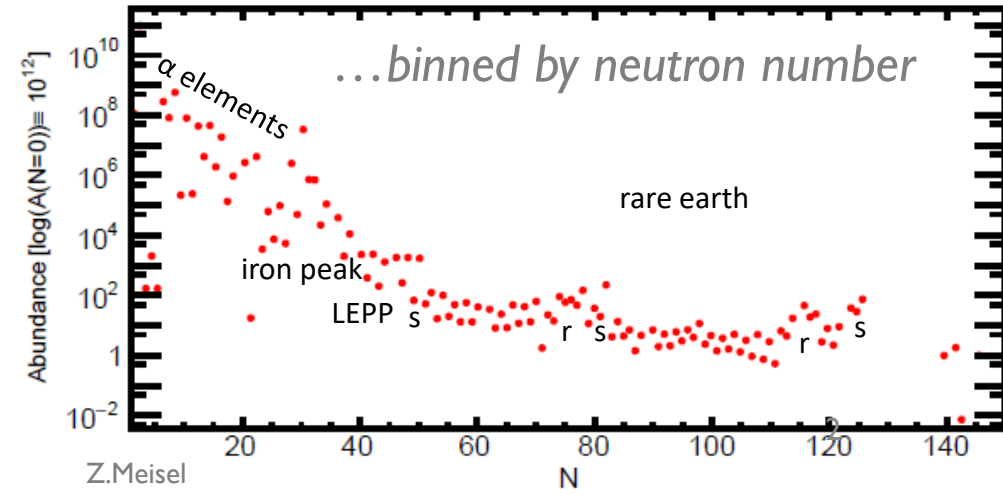
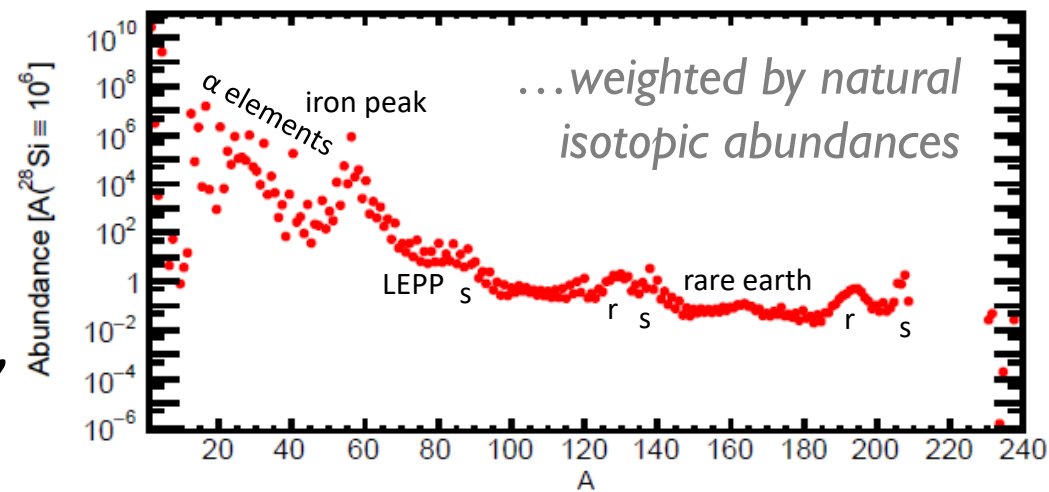
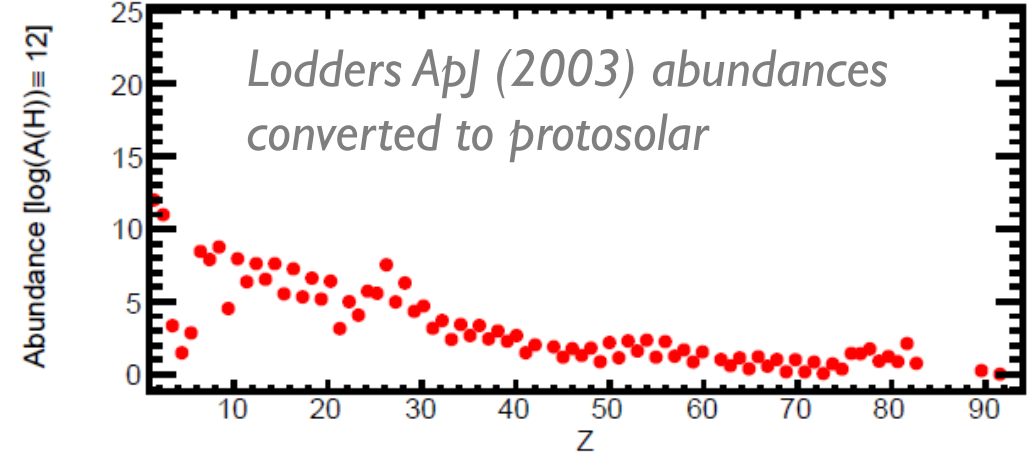
Quick notes on
s-process

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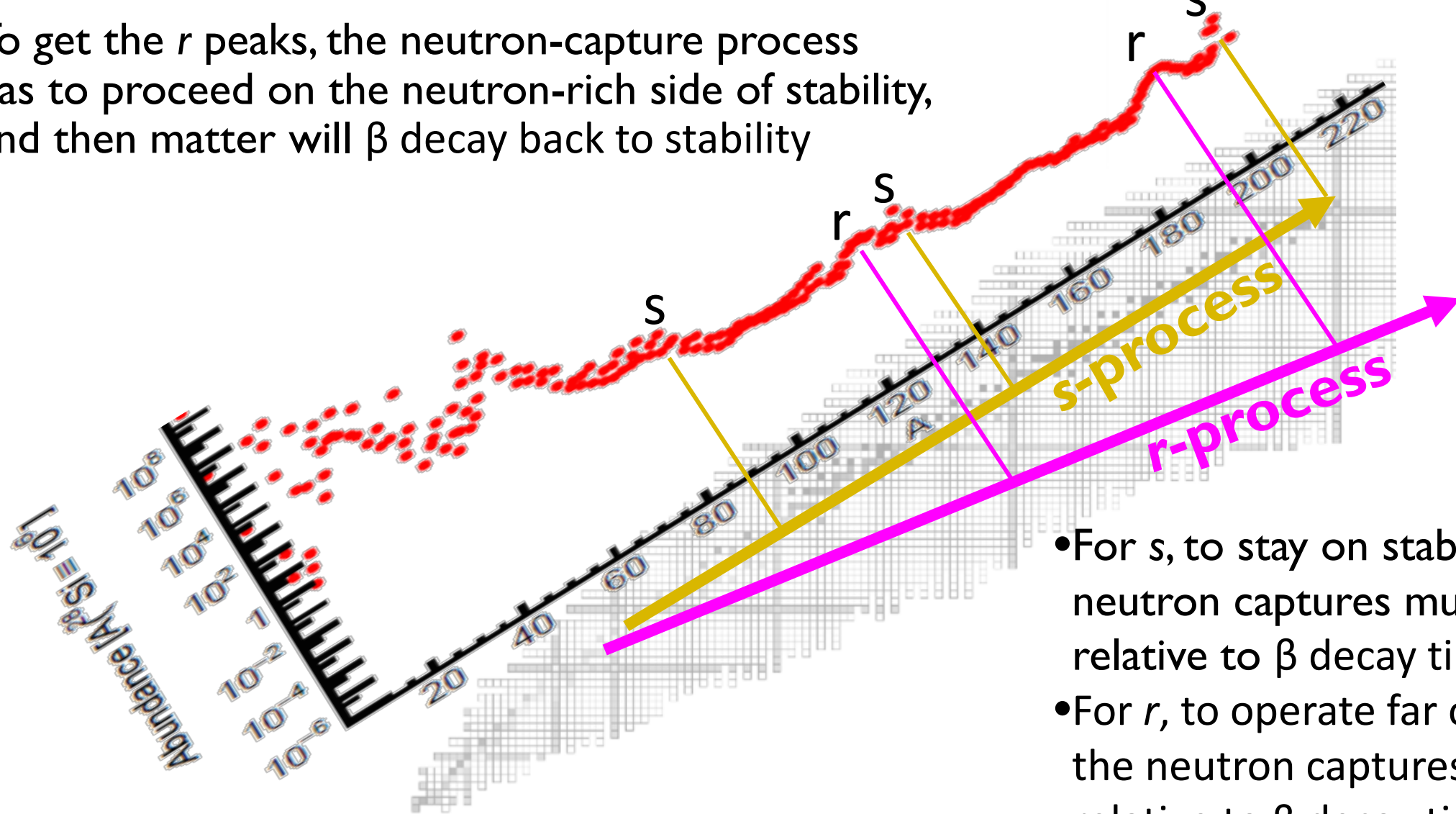
The case for neutron-capture processes

- Looking at the solar system abundances in terms of A , we see some pretty interesting patterns
- The α elements and iron peak are consistent with our picture for massive star evolution, but we see several other features beyond iron
- Focusing on the peaks labeled s and r , we see that the s peaks are located at neutron magic numbers 50, 82, 126, with the r peaks located just below (at least for 82 & 126)
- Logically, it follows that the processes making the s and r processes somehow involve sequences of neutron capture reactions that pass through these magic numbers on the nuclear chart. This is because one feature of magic N nuclei is low neutron-capture cross sections relative to neighboring nuclei



We can see where on the nuclear chart the *s* & *r* processes must have occurred by considering where in terms of Z pile-up at a magic N must have occurred to give the observed peak in A

- A neutron-capture process proceeding along stability provides *s* peaks
- To get the *r* peaks, the neutron-capture process has to proceed on the neutron-rich side of stability, and then matter will β decay back to stability



- For *s*, to stay on stability, the neutron captures must be slow relative to β decay timescales
- For *r*, to operate far off stability, the neutron captures must be rapid relative to β decay timescales³

Neutron-capture time-scale: 's'-process

Time-scale for 'slow' neutron capture

1. $\sim 10^8$ neutrons in a box of 1 cm^3 , each moving from thermal velocity
2. $\sim 1/6$ leave a single box side with a velocity given by:

- $E \sim kT \sim mv^2 \rightarrow v = \sqrt{\frac{kT}{m}} \approx \sqrt{\frac{25\text{keV}}{10^{-27}\text{kg}}} \approx 10^9\text{cm/sec}$

3. $1/6$ of the neutrons leave from a cube face, with 1 cm^2 area every 10^{-9} seconds

- \rightarrow neutron flux $\sim 10^{17}$ neutrons/cm²/sec

4. neutron-capture cross-section (at $\sim 25\text{keV}$): $\sim 100\text{mb} = 10^{-25}\text{cm}^2$

5. neutron-capture rate = (flux)x(cross-section)
 $\approx 10^{17}$ neutrons/cm²/sec $\times 10^{-25}$ cm² = 10^{-8} /sec

6. neutron-capture time = $1/\text{Rate} \approx 10^8$ seconds ~ 1 decade

s-process nucleosynthesis

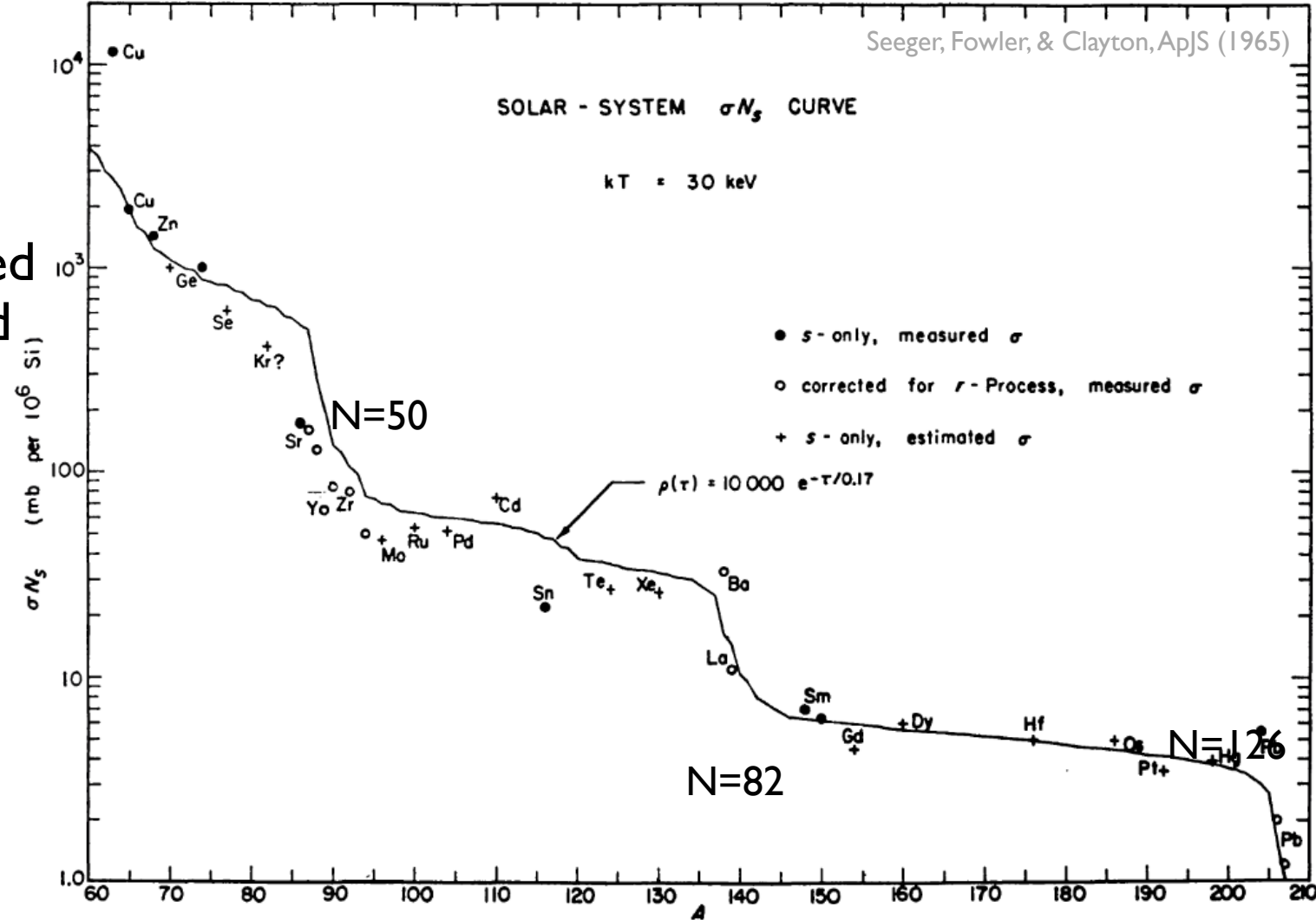
- For the s-process, we need an environment where nuclei can bathe for a long time in a moderate neutron density ($n_n \sim 10^7 - 10^{10} \text{ cm}^{-3}$). Let's consider the reaction flow for this case.
- For isotope (Z,A), the abundance change is the sum of production & destruction mechanisms,
$$\frac{dN_{Z,A}}{dt}(t) = n_n(t)n_{Z,A-1}(t)\langle\sigma v\rangle_{Z,A-1} - n_n(t)n_{Z,A}(t)\langle\sigma v\rangle_{Z,A} - \lambda_{\beta:Z,A}n_{Z,A}(t)$$
- We'll ignore species with fast β -decays, since they'll just form the more stable isobar more or less instantly, so we can set the last term to zero
- Note that $\langle\sigma v\rangle_{n \text{ cap}} \sim \text{constant}$, and anyhow we'll assume a constant temperature, so $\langle\sigma v\rangle_i \rightarrow \sigma_i v_T$, where v_T is the thermal velocity for the environment temperature
- As such, our abundance change equation is now
$$\frac{dN_{Z,A}}{dt}(t) = n_n(t)v_T(\sigma_{Z,A-1}n_{Z,A-1} - \sigma_{Z,A}n_{Z,A})$$
- Note that $n_n(t)v_T$ is the neutron flux. Integrating this over time gives the neutron irradiation $\tau = \int_0^t n_n(t)v_T dt = v_T \int_0^t n_n(t) dt$, which is referred to as the neutron exposure
- Re-casting our abundance change in terms of τ ,
$$\frac{dN_{Z,A}}{d\tau} = \sigma_{Z,A-1}n_{Z,A-1} - \sigma_{Z,A}n_{Z,A}$$
- In equilibrium, $\frac{dN_{Z,A}}{d\tau} = 0$, so $\sigma_{Z,A-1}n_{Z,A-1} = \sigma_{Z,A}n_{Z,A} = \text{constant}$
- So, we can get s-process relative abundances based solely on neutron-capture cross sections!

The steady flow approximation

- The steady-flow approximation holds roughly in between magic N
- Tests can be performed for elements with multiple isotopes that are shielded from the r -process and have measured neutron-capture cross sections, e.g. $^{122,123,124}\text{Te}$

^{122}Te	^{123}Te	^{124}Te	^{125}Te	^{126}Te	^{127}Te
^{121}Sb	^{122}Sb	^{123}Sb	^{124}Sb	^{125}Sb	^{126}Sb
^{120}Sn	^{121}Sn	^{122}Sn	^{123}Sn	^{124}Sn	^{125}Sn
^{119}In	^{120}In	^{121}In	^{122}In	^{123}In	^{124}In

r



Reactions of special interest are those with half-lives similar to the neutron-capture time. They're called branch-points. They not only alter the abundance pattern, but can be used as s -process thermometers and neutron-density probes, since $\langle\sigma v\rangle$ isn't exactly constant.

Where do the s neutrons come from?

- More careful analyses (e.g. checks with the Te isotopes and branch-points) demonstrate the need for different neutron fluxes for different durations.

These are the

- **Weak:** $60 < A < 90$, near He/C shells in massive stars
- **Main:** Sr-Pb, thermal pluses in AGB stars
- **Strong:** Pb enhancement for low-metal AGB stars

components. For each $kT \sim 30\text{keV}$ or thereabouts.

- No matter, the neutron sources turn-out to be $^{13}\text{C}(\alpha, n)$ and $^{22}\text{Ne}(\alpha, n)$, where the former only operates for the AGB cases and the latter only operates during brief flashes for the AGB cases

